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nucleus. ${}^1_0\text{n}$. Particle for Neutron Ejection. ${}^5_2\text{He} \rightarrow {}^1_0\text{n} + {}^4_2\text{He}$.

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Chapter 23 Nuclear Chemistry Notes 1

CHAPTER 23 NUCLEAR CHEMISTRY 23.1

THE NATURE OF NUCLEAR REACTIONS

radioactivity - the spontaneous decay of an unstable nucleus with accompanying emission of radiation. nuclide - atom with a specific number of protons and

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neutrons in its nucleus. \Rightarrow There are 271 stable nuclides in nature, others are radioactive

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25.5 = Applications and Effects of Nuclear Reactions (Day 17 = A-5/8, A-5/11) Chapter 25 Homework (Due Day 18 = A-5/12, B-5/13)

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Radioactivity • Radioactivity is the process by which nuclei emit particles and rays as they break down. • The name of the penetrating rays emitted by

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a radioactive source is called radiation.

- A radioactive isotope is an unstable atom which breaks down on its own, releasing energy and/or

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10.E: Nuclear and Chemical Reactions
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exercises to accompany Chapter 10 of the University of Kentucky's LibreText for CHE 103 - Chemistry for Allied Health. Solutions are available below the questions.

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